

TECHNICAL REPORTS

Surface Water Quality

Variability in fluvial suspended and streambed sediment phosphorus fractions among small agricultural streams

Sara Sandström¹  | Martyn N. Futter¹ | David W. O'Connell² |
 Emma E. Lannergård¹ | Jelena Rakovic³  | Katarina Kyllmar³ | Laurence W. Gill² |
 Faruk Djodjic¹ 

¹ Dep. of Aquatic Sciences and Assessment, Swedish Univ. of Agricultural Sciences, P.O. Box 7050, Uppsala, Sweden

² Dep. of Civil and Environmental Engineering, Museum Building, Trinity College Dublin, the University of Dublin, Dublin 2, Ireland

³ Dep. of Soil and Environment, Swedish Univ. of Agricultural Sciences, P.O. Box 7014, Uppsala, Sweden

Correspondence

Sara Sandström, Dep. of Aquatic Sciences and Assessment, Swedish Univ. of Agricultural Sciences, P.O. Box 7050, Uppsala, Sweden.

Email: sara.sandstrom@slu.se

Assigned to Associate Editor Jian Liu.

Abstract

Agriculture is a major source of sediment and particulate phosphorus (P) inputs to freshwaters. Distinguishing between P fractions in sediment can aid in understanding its eutrophication risk. Although streams and rivers are important parts of the P cycle in agricultural catchments, streambed sediment and especially fluvial suspended sediment (FSS) and its P fractions are less studied. To address this knowledge gap, seasonal variations in FSS P fractions and their relation to water quality and streambed sediment were examined in three Swedish agricultural headwater catchments over 2 yr. Sequential fractionation was used to characterize P fractions in both streambed sediment and FSS. All catchments had similar annual P losses (0.4–0.8 kg ha⁻¹), suspended solids (124–183 mg L⁻¹), and FSS total P concentrations (1.15–1.19 mg g⁻¹). However, distribution of P fractions and the dominant P fractions in FSS differed among catchments ($p < .05$), which was most likely dependent on differences in catchment geology, clay content, external P sources, and flow conditions. The most prominent seasonal pattern in all catchments was found for iron-bound P, with high concentrations during low summer flows and low concentrations during winter high flows. Streambed sediment P fractions were in the same concentration ranges as in FSS, and the distribution of the fractions differed between catchments. This study highlights the need to quantify P fractions, not just total P in FSS, to obtain a more complete understanding of the eutrophication risk posed by agricultural sediment losses.

1 | INTRODUCTION

Phosphorus (P) is a primary limiting nutrient in freshwater ecosystems (Correll, 1998; Schindler, 1974). Agricultural

activities generate spatially variable and episodic P inputs to surface waters (Sharpley et al., 2009). Such nonstationary dynamics can lead to challenges for understanding the primary processes governing P mobility in different catchments as well as difficulties in predicting and implementing appropriate mitigation policies and countermeasures. These issues are common in agricultural headwater catchments, which can have a considerable influence on downstream rivers and lakes (Bol et al., 2018).

Abbreviations: Al-P, aluminum-bound phosphorus; Ca-P, calcium-bound phosphorus; Fe-P, iron-bound phosphorus; FFS, fluvial suspended sediment; H₂O-P, loosely bound phosphorus; OM, organic matter; org-P, organically bound phosphorus; PP, particulate phosphorus; RP, reactive phosphorus; SS, suspended solids; TP, total phosphorus.

This is an open access article under the terms of the [Creative Commons Attribution-NonCommercial](https://creativecommons.org/licenses/by-nc/4.0/) License, which permits use, distribution and reproduction in any medium, provided the original work is properly cited and is not used for commercial purposes.

© 2021 The Authors. *Journal of Environmental Quality* published by Wiley Periodicals LLC on behalf of American Society of Agronomy, Crop Science Society of America, and Soil Science Society of America

Historical land management practices have caused accumulation of legacy P in soils and freshwater sediments (Kleinman et al., 2011; Sharpley et al., 2013) where biogeochemical processes can transform recalcitrant P fractions to more bioavailable forms (Lannergård et al., 2020). Although streambed sediments are important sources and sinks for P cycling (Ezzati et al., 2020; Jarvie et al., 2005; Owens & Walling, 2002; Weigelhofer et al., 2018), the role of fluvial suspended sediment (FSS) in P cycling is rarely investigated despite its potential significance. Different P fractions (e.g., phosphate-P, organic P, and metal-bound P) differ in mobility and bioavailability, which governs impacts on water quality and aquatic ecosystem functioning (Kaiserli et al., 2002; Ye et al., 2017). New insights into the relative and absolute amounts of different P fractions in both streambed and FSS will support more comprehensive assessments of potential effects of agriculture on aquatic ecosystems and generate more focused and effective catchment-specific eutrophication mitigation strategies.

Several chemical sequential extraction schemes exist to quantify different P fractions in sediment (Goedkoop & Pettersson, 2000; Hupfer et al., 2009; Psenner, 1988; Ruttensberg et al., 2009; Williams et al., 1976). Most schemes define operational P fractions, including loosely sorbed P, metal-associated P (e.g., iron [Fe] and aluminum [Al]), and calcium (Ca)-bound P (Condrón & Newman, 2011). Numerous studies exist on P fractions in lake sediment (Agstam-Norlin et al., 2020; Kaiserli et al., 2002; O'Connell et al., 2020; Pettersson, 1998), and there are a few studies of P fractions in streambed sediment (Audette et al., 2018; Lannergård et al., 2020; San-Clements et al., 2009). However, studies of FSS are rare (Balandine et al., 2008; Nguyen & Sukias, 2002), and knowledge about seasonal variations is needed.

The study aims were to quantify patterns in the distribution of P fractions in FSS and streambed sediment in three headwater agricultural catchments and investigate potential controlling factors (Aim 1), to investigate seasonal changes in distribution of FSS P fractions (Aim 2), and to explore relationships between water column phosphate P and the labile FSS P fraction as well as between water column and FSS total P (TP) (Aim 3). To address the study aims, FSS and streambed sediment from three well-monitored agricultural headwater catchments in southern Sweden were sampled over 2 yr and analyzed for different P fractions.

2 | MATERIALS AND METHODS

2.1 | Catchment descriptions

Three small southern Swedish agricultural catchments (C6, U8, and E23) in the Swedish Environmental Monitoring program (Kyllmar et al., 2014) with a history of high P load and suspended solids (SS) transport were selected (Figure 1;

Core Ideas

- Similar total phosphorus losses were seen in three agricultural headwater streams.
- P fractions in suspended and bed sediment varied significantly among catchments.
- Large seasonal variations in suspended sediment P fractions were seen.
- Dominant P fractions are dependent on clay content, geology, and flow.
- Suspended and bed sediment P fractions should be included in routine monitoring.

Tables 1 and 2) (Sandström et al., 2020). All three catchments are dominated by agricultural land (>50%) and clay-rich soils. The smallest catchment, U8, has the heaviest soil texture (Table 1). Streams in U8 and E23 are small and can experience extremely low flows, especially in summer. The most southern catchment, E23, has a markedly higher stocking density and higher rates of manure applications (Table 1), whereas C6 has the lowest rate (Linefur et al., 2019). The U8 outlet is surrounded by agricultural fields, and the stream channel can become overgrown with vegetation during summer. The C6 stream outlet is also surrounded by agricultural fields, and alongside the outlet there is a wetland-like terrace that is inundated during high flows. Streambed sediments at the outlet of both U8 and C6 consist mainly of clay, overlain by organic material (2–5 cm). In E23, the catchment outlet runs through a forested area, with stones and twigs in the stream. Streambed sediment in E23 is harder than in U8 and C6 and contains substantially more gravel and sand.

2.2 | Fluvial suspended sediment and streambed sediment: Sampling strategy and sample preparation

Two Phillips samplers (Supplemental Figure S1) (Phillips et al., 2000) were placed at the outlet of each catchment to collect FSS. The samplers were positioned mid-stream at approximately mean depth above the streambed, with the inlet facing upstream (Phillips et al., 2000). During the first year, samplers were emptied approximately every fourth week into 10-L buckets (to collect FSS for further analyses) and then put back into the stream (Figure 1). Sampling started in C6 and U8 during autumn 2017, and E23 was added in autumn 2018. During 2018/2019, samplers were emptied every 8–10 wk to ensure enough FSS was collected for later analyses. From October 2017 to May 2018, the C6 stream was flooded, and samplers were not retrievable during this period; hence,

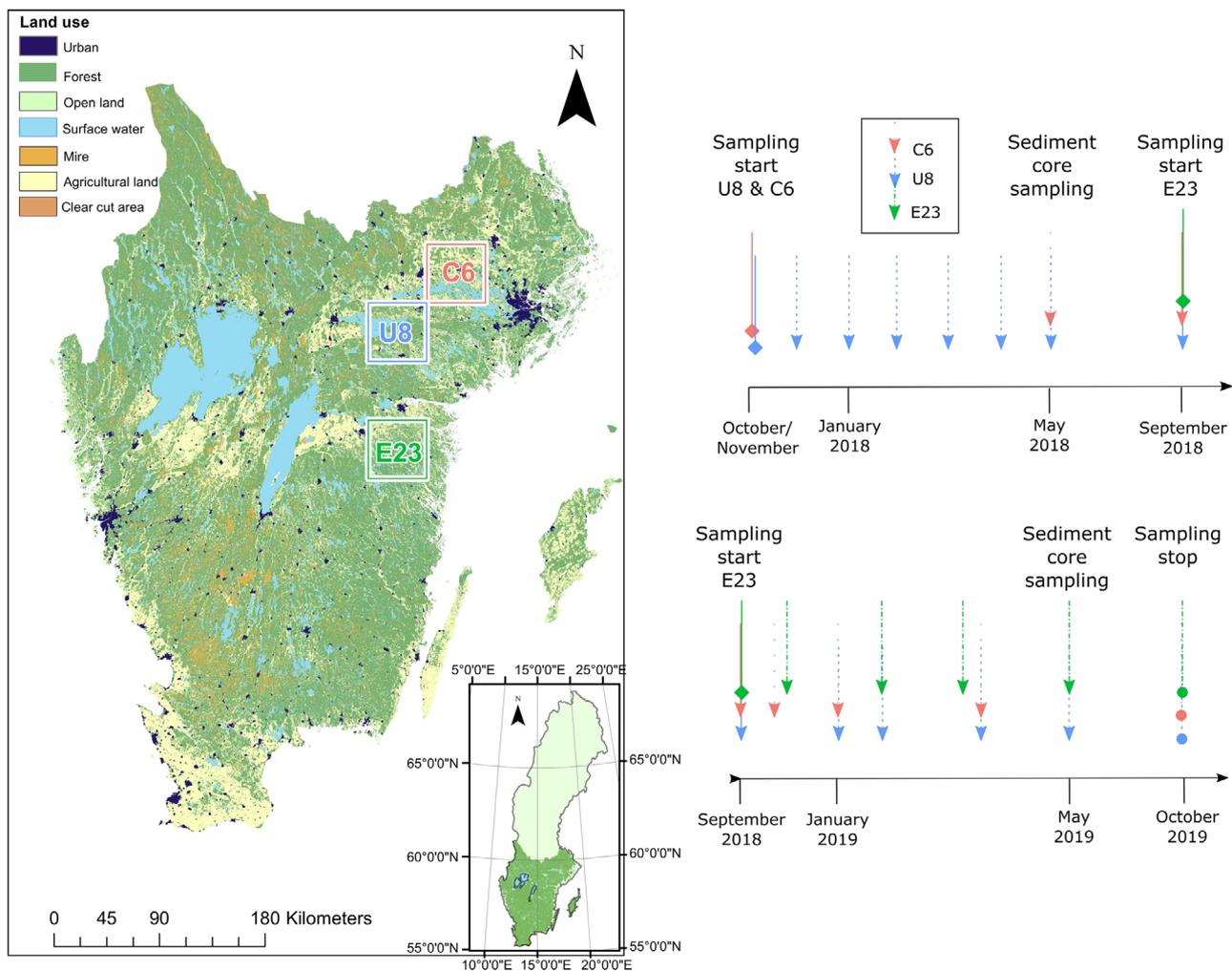


FIGURE 1 (Left) Locations of the three study catchments in southern Sweden including land use as the background map. Exact locations cannot be provided due to agreement with the farmers. The inset map displays Sweden with marked latitude and longitude lines. (Right) Illustration of sampling scheme of fluvial sediment and streambed sediment. The different colors represent the different catchments. Rhombi indicate sampling start for each individual catchment, filled circles represent the time at which sampling stopped, and colored arrows indicate sampling dates

one sample represents this entire 6-mo period. From April to May 2019 there was not enough material in the C6 samplers for analysis. All samples were brought back to the laboratory and wet sieved through a 0.063-mm nylon mesh to collect the smallest fractions and remove plant material. Samples were stored in 2-L buckets at 4 °C for at least 24 h to allow particles to settle. After sedimentation, as much supernatant as possible was removed through decanting and pipetting. The remaining material was transferred to a 50-cl centrifuge tube and centrifuged for 5 min at 3,500 rpm, and the supernatant was removed by decanting. Samples were stored in airtight boxes at 4 °C until analysis.

Streambed sediment cores were taken in each catchment at a mid-channel location as close as possible upstream from the Phillips samplers to get a representative sample from the stream rather than the streambanks. Each core was collected using a Willner gravity corer on a rod or by simply pushing

the tube in manually, with two plugs creating a vacuum. In U8 and C6, cores were collected in May 2018. The E23 core was collected in May 2019. Cores were collected after the spring flood and before too much vegetation had grown to avoid plant disturbance. Due to the coarse and hard E23 streambed, this core was markedly shorter (4 cm) than cores from the other catchments (8–9 cm). All cores were sliced into 1 cm (where possible) slices in the field. Slices were stored in airtight cans at 4 °C until analysis (slices were held for a maximum of 6 wk).

2.3 | Water quality sampling and analysis

In each catchment, daily discharge was measured at the stream outlet by continuous water level recording at well-defined sections (i.e., a v-notch or broad-crested weir) (Kyllmar et al.,

TABLE 1 Catchment characteristics

Catchment ID	Sampling period	Area km ²	Arable land %	Stocking density ^a AU ha ⁻¹	Scattered households persons km ⁻²	Applied manure P (mean 2007–2010) kg ha ⁻¹	Three dominant soil texture classes ^b	Dominating bedrock (>50%) (main mineralogy ^c)	Precipitation ^d mm
C6	Oct. 2017–Oct. 2019	33.1	59	<0.1	10	0.35	silty clay (77%), clay loam (14%), clay (4%)	quartz arenite (feldspar-mica-granite)	521
E23	Sept. 2018–Oct. 2019	7.4	54	0.6	7	17	clay loam (41%), silty clay (31%), clay (19%)	granite (feldspar-mica-granite)	591
U8	Oct. 2017–Oct. 2019	5.7	56	<0.1	11	8	silty clay (85%), clay (14%), clay loam (1%)	granodiorite-granite (feldspar-mica-granite)	539

^aStocking density for all catchments, amount of P applied to E23, U8 (Kyllmar et al., 2014), and C6 (Linefur et al., 2017); 1 AU refers to one cow, three pigs, 10 sheep, or 100 hens.

^bUSDA data from Söderström and Piikki (2016).

^cBedrock information is from Geological Survey of Sweden (SGU, 2016).

^dPrecipitation and temperature are normal values (i.e., a long-term mean value, from SMHI 1961–1990; Alexandersson et al., 2001).

2014). Automatic flow-proportional and grab water samples were collected at the catchments' outlets within the frames of the Swedish Environmental Monitoring Program (Kyllmar et al., 2014). All samples were analyzed following Swedish Standard Methods (Kyllmar et al., 2014). Parameters reported here include total P (TP), phosphate-P (reactive P [RP], following Haygarth & Sharpley [2000]), and SS. Total P was analyzed on unfiltered water samples after digestion, whereas RP was analyzed after filtration at 0.2 μm , both using the molybdate blue method (Murphy & Riley, 1962). Particulate P (PP) was estimated as the difference between TP and RP. Estimates of flow-weighted concentrations and loads were based on Linefur et al. (2019) (see Methods in Supplemental Material).

2.4 | Sequential P fractionation

To analyze samples (suspended and streambed sediment) for P concentration and different P fractions, a sequential P fractionation method based on Psenner and Pucsko (1988), Psenner et al. (1984), and Hupfer et al. (1995), Hupfer, Zak et al. (2009) was used. The P fractions include loosely sorbed P, redox-sensitive P (iron/manganese [Fe/Mn] bound P), nonreducible and Al-hydroxide-bound P, organically bound P (org-P), Ca bound P (Ca-P), and refractory P. The main difference between the method used here and the aforementioned studies is omission of the refractory P step. All sediment samples were analyzed fresh and in triplicate. All fractions are operationally defined and will be referred to by their operational name as follows in parenthesis: extraction with Milli-Q water (H₂O-P), buffered dithionate solution (Fe-P), and NaOH (Al-P), organically bound P (org-P, defined as the difference between Al-P and extraction with NaOH following digestion), and extraction with HCl (Ca-P). All samples were analyzed at the end for RP (unfiltered) following Murphy and Riley (1962). All fractionation results are presented as mg g⁻¹ in the context of FSS and streambed sediment comparisons or as mg L⁻¹ for comparison to water column concentrations. The sum of all P fractions is hereafter termed "TP" because it has been found to be within $\pm 10\%$ of total sediment P (Huser & Rydin, 2005; Psenner, 1988). The first two steps in the sequential P fractionation method (H₂O-P and Fe-P) are commonly referred to as labile P (Huser et al., 2016; Reitzel et al., 2005, 2006) and were used for comparisons with RP in water columns.

All sediment samples were analyzed for dry matter content by freezing for at least 12 h at $-20\text{ }^{\circ}\text{C}$ and then freeze drying for at least 4 d. Loss on ignition to determine organic matter (OM) content of the sample was performed by igniting a freeze-dried sample for 2 h at $550\text{ }^{\circ}\text{C}$ (Håkanson & Jansson, 1983).

TABLE 2 Catchment data

Catchment ID	Temp.	pH (H ₂ O) ^a	TP conc. ^b	RP conc. ^b	PP conc. ^b	SS conc. ^b	TP load ^c	Discharge ^d
	°C							
C6	5.5	7.5	0.21	0.05	0.15	144	0.4	226
E23	6.3	7.6	0.28	0.12	0.14	124	0.4	176
U8	5.9	7.3	0.26	0.06	0.19	183	0.8	230

^aThe pH value represents a mean value for the study period.

^bLong-term mean concentrations of total P (TP), phosphate P (RP), particulate P (PP), and suspended solids (SS) (U8 and E23: 2007–2017; C6: 2004–2017) (Sandström et al., 2020).

^cLong-term average load of TP (U8 and E23: 1995/1996–2014/2015; C6: 2005/2006–2014/2015) (Linefur et al., 2017).

^dLong-term (1995/1996–2016/2017) average discharge (Linefur et al., 2019).

2.5 | Statistical analysis

Two-sided Wilcoxon rank sum tests were performed, where a p value $< .05$ was defined as significant. The same significance threshold was used for linear regressions and correlations. All sequential P fractionation results are presented as a mean of three subsamples. Standard deviations for each result are reported (Supplemental Tables S1 and S2).

To compare streambed sediment P fractions to concentrations in FSS, fluvial samples collected immediately prior to sediment core collections were used. Concentrations of P fractions in the top 4 cm of sediment were used because this depth is generally considered to be active (Reynolds & Davies, 2001). All statistical analyses were performed using R 3.6.2 (R Core Team, 2019).

3 | RESULTS

3.1 | Seasonal and between-catchment variations in FSS P fractions

Small temporal variations were observed for H₂O-P in U8 and E23 (Figure 2a). Proportions of H₂O-P were low and similar for all catchments (3.6–6.6% of TP; Figure 3a). The most prominent temporal change in concentration was recorded for the Fe-P fraction (Figure 2b), with a similar pattern in all catchments. The highest Fe-P concentrations (0.40–0.52 mg g⁻¹) were observed during summer; the lowest concentrations (0.13–0.23 mg g⁻¹) were observed during winter and spring (Figure 2b; Supplemental Table S1). Variation in Fe-P concentration was highest in C6 (CV, 56%; Supplemental Table S1), where it was the dominant fraction (Figure 3b).

In U8, Al-P was the dominant fraction (36%) and was significantly ($p < .005$) higher than in the other catchments (Figures 2c and 3c). For C6, the Al-P fraction was relatively stable over the study period (Figure 2c), with low total variation (0.16–0.26 mg g⁻¹; 11–27%) (Figure 3c). In E23 there seemed to be a monotonic decrease of Al-P with low total variation (0.09–0.30 mg g⁻¹; 10–16%) (Figure 3c).

Patterns for org-P were similar for all catchments, except for one peak in E23 in autumn 2018 (Figure 2d). In U8 and E23, the org-P fraction had the highest concentration variation among the different fractions (CV, 68 and 76%, respectively) (Figure 3d; Supplemental Table S2). In E23, org-P was the dominant fraction (32%) (Figures 2d and 3d). Organically bound P and Fe-P in FSS correlated positively ($R^2 = .37$, $n = 23$, $p < .05$). No other significant correlations between FSS P fractions were observed.

There was a significant difference between U8 and both C6 and E23 for the Ca-P fraction. However, the difference for Ca-P was not as clear as for Al-P (Figure 3e). Calcium-bound P was otherwise fairly stable for all catchments and had low total variation in concentrations (U8: CV, 30%; C6: CV, 6%, E23: CV, 23%) (Figures 2e and 3e; Supplemental Table S1), except for one peak in E23 and U8. Variation in TP concentrations was similar between catchments and had a clear seasonal pattern, with higher concentrations during summer and early autumn and a decrease during late autumn and winter (Figure 2f).

For U8, fractional H₂O-P concentrations were significantly lower ($p < .001$) than all other fractions, and Fe-P was significantly lower than the Al-P fraction, which was significantly higher than both the org-P and Ca-P fractions. At C6, the H₂O-P fraction was significantly lower ($p < .005$) than all other fractions. At E23, the H₂O-P fraction was significantly lower than all other fractions except for Al-P, which was significantly lower than the org-P fraction.

3.2 | Relations between water quality data, discharge, and FSS P fractions

Flow-weighted TP and SS concentrations correlated well when all catchments were used in the same linear regression ($R^2 = .71$, $n = 24$, $p < .001$). When tested separately, TP and SS correlated in U8 ($R^2 = .83$, $n = 12$, $p < .001$) and C6 ($R^2 = .90$, $n = 7$, $p < .01$) but not in E23 ($R^2 = .11$, $n = 5$, $p > .5$). The correlation between PP

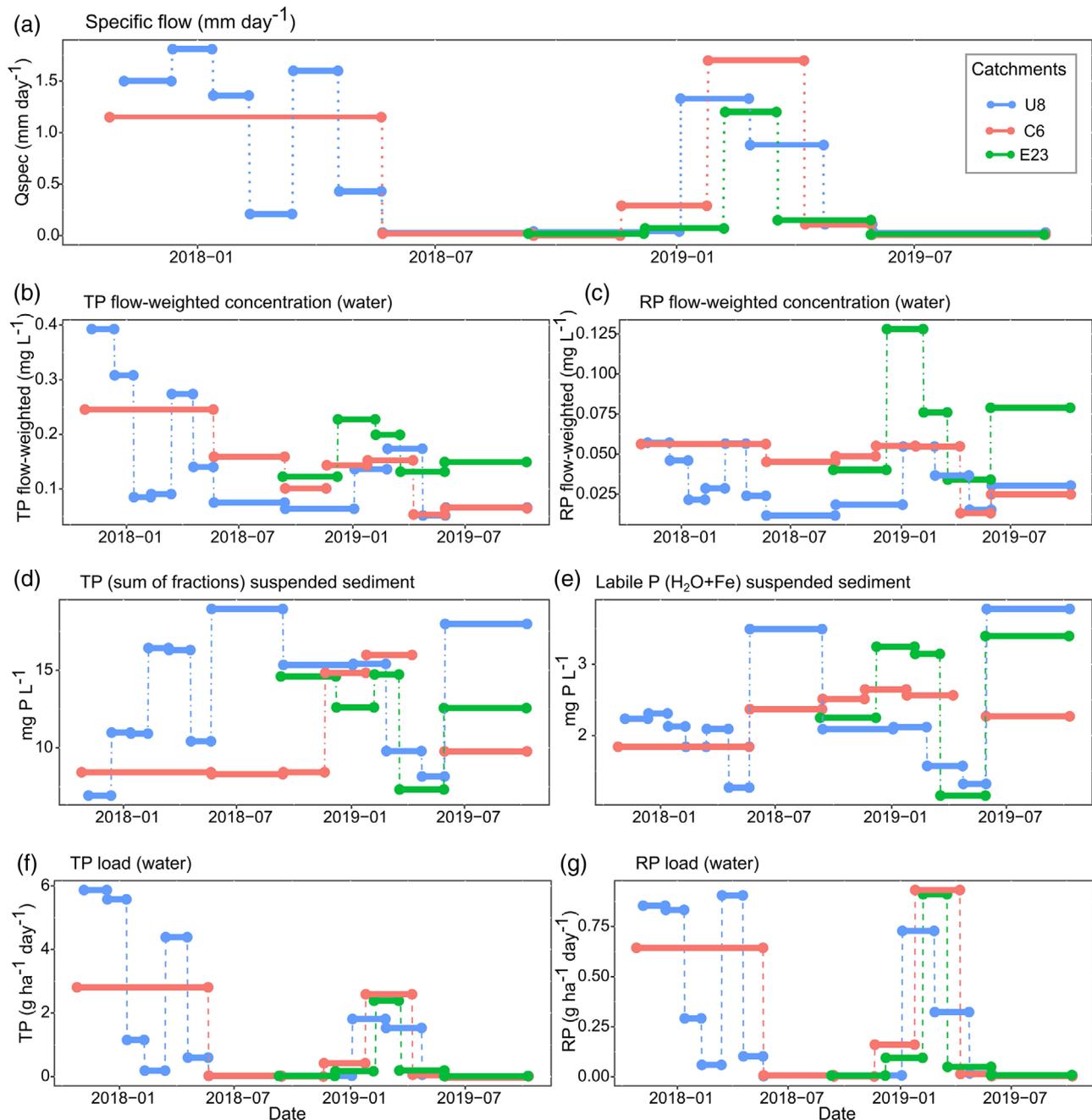


FIGURE 2 (a–e, g) Seasonal variations of the different phosphorus (P) fractions (Al-P, aluminum-bound P; Ca-P, calcium-bound P; Fe-P, iron bound P; H₂O-P, loosely bound P; org-P, organically bound P; TP, total P) over the sampling period in the three study catchments. Each bar in the graphs represents the concentration of the composite sample from the period the sampler was in the stream; different colors represent different catchments. The dotted lines are inserted to make it easier to differentiate the sampling periods. Sampling in E23 started approximately 1 yr after the other catchments, in September 2018 (green bars and dotted lines). Between April and May 2019 in catchment C6, not enough material was collected to perform the analysis, hence the tilted dotted red line. (f) Sum of the different fractions. Note that the y-axes are differently scaled for each fraction to enhance visibility of the individual changes of each fraction

and SS flow-weighted concentrations when all catchments were used in the same linear regression analysis was even higher ($R^2 = .80$, $n = 24$, $p < .001$) (Supplemental Figure S2). The only significant linear correlation between flow-weighted RP and SS concentrations was at U8 ($R^2 = .43$, $n = 12$, $p < .05$).

Annual specific discharge was well below average in all catchments in the second study year (Table 2; Supplemental Table S2; Supplemental Figure S3a). In all catchments, TP and RP loads followed the same pattern as the specific discharge (Supplemental Figure S3a,f,g). All catchments have similar flow-weighted TP concentrations, especially during

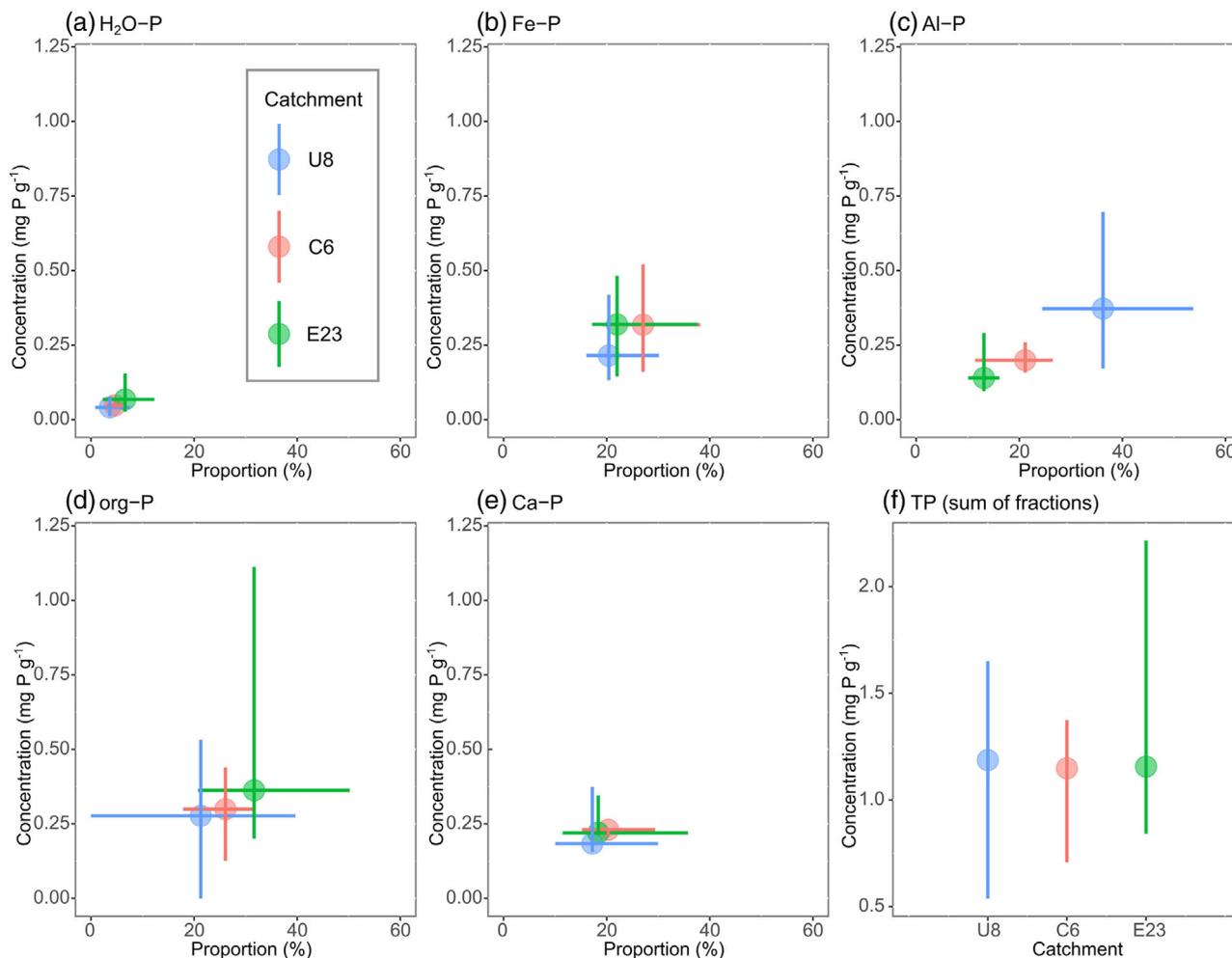


FIGURE 3 (a–e) Total variation of each phosphorus (P) fraction (Al-P, aluminum-bound P; Ca-P, calcium-bound P; Fe-P, iron bound P; H₂O-P, loosely bound P; org-P, organically bound P; TP, total P) in the fluvial suspended sediment and the sum of all fractions (f), colored by catchment. Dots represent the median value of each fraction, and the ends of the lines represent the minimum and the maximum values, respectively. The x-axis represents percentage of fraction, and the y-axis shows concentration of the fraction. In (f), only the variation in concentration of the sum of the fractions is displayed

the last study year (Supplemental Figure S3b). Flow-weighted water column and FSS TP concentrations showed no significant correlation (data not shown).

Flow-weighted water column RP concentrations seemed to follow a similar temporal pattern as FSS labile P (H₂O-P + Fe-P) (Supplemental Figure S3c,e). In C6 and E23, especially, a significant but weak correlation existed between the two variables ($R^2 = .16$, $n = 24$, $p < .05$) (Supplemental Figure S4). The RP/TP ratio was generally low for all catchments over the entire study period (Supplemental Table S2).

3.3 | Comparison of P fractions in FSS and the streambed active sediment layer

In all catchments, concentrations of different FSS P fractions immediately prior to sediment core collections generally

followed concentrations in active streambed sediments (Figure 4) and were in the same concentration ranges. However, there were some clear differences.

In C6, Ca-P was higher in streambed sediment (64%) than in FSS (29%), where Ca-P was also the largest fraction (Figure 4a; Supplemental Figure S5). There were similar proportions (18 and 23%, respectively; Supplemental Figure S5) of the Fe-P fraction in the top streambed sediment and FSS. In streambed sediment, org-P was the smallest fraction, whereas H₂O-P was the smallest fraction in FSS (Figure 4a; Supplemental Figure S5).

In U8 streambed sediment, org-P, Al-P, and Fe-P concentrations and fractions were similar at 27, 30, and 30%, respectively (Figure 4b; Supplemental Figure S5). Relative fractions in FSS and streambed sediment were similar, except that org-P was the dominant fraction in FSS (40%) (Supplemental Figure S5).

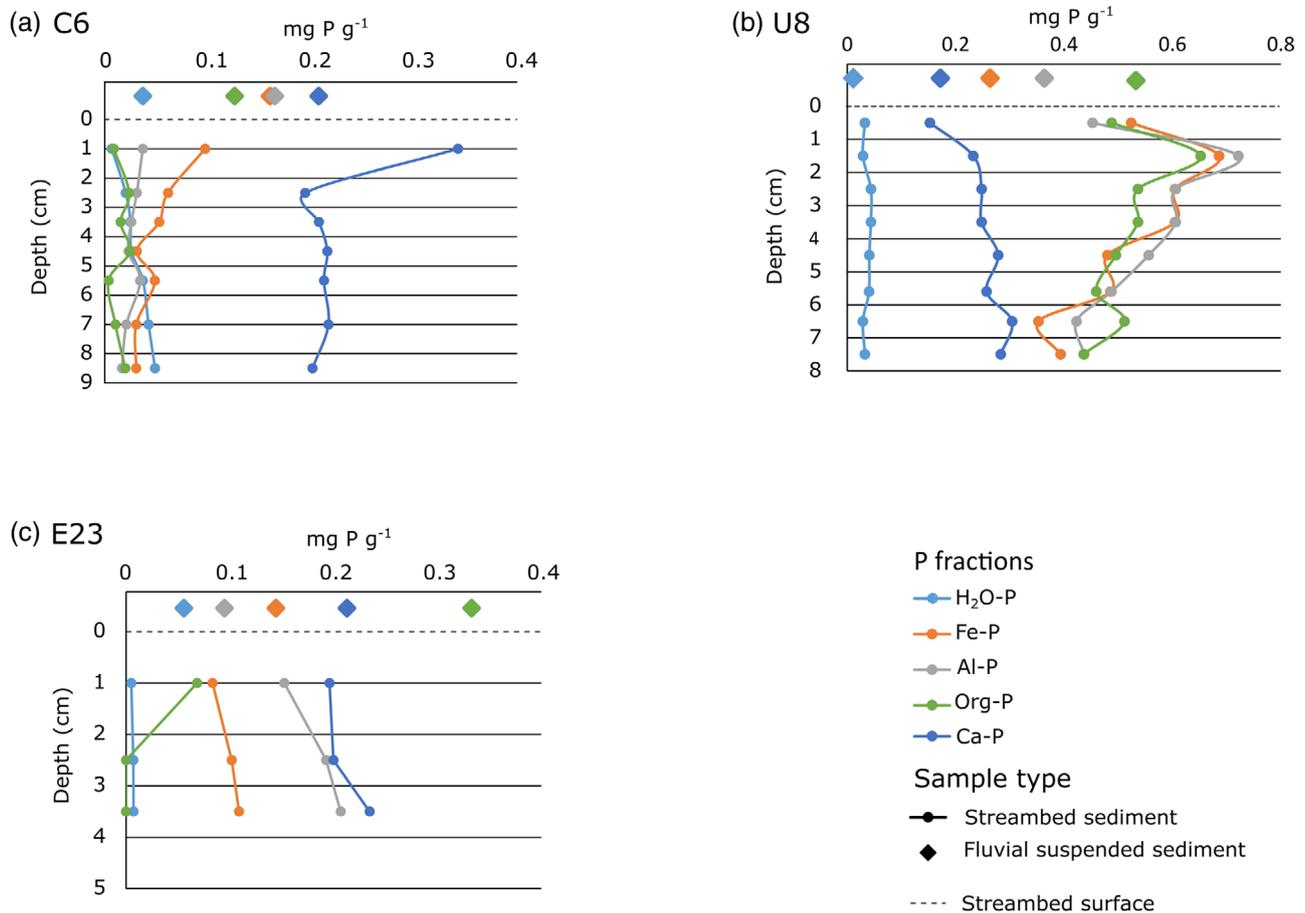


FIGURE 4 Concentrations of phosphorus (P) fractions (Al-P, aluminum-bound P; Ca-P, calcium-bound P; Fe-P, iron bound P; H₂O-P, loosely bound P; org-P, organically bound P; TP, total P) in sediment cores from the three catchments (dots and lines) and in fluvial suspended sediment (rhombi) from the period preceding sediment core collection. The different colors represent the different fractions, and a–c represent the different catchments. Each value is a mean value of triplicate samples from the same sediment sample

In E23, concentrations and fractions of org-P in FSS were clearly higher than in streambed sediment (Figure 4c), where Ca-P was dominant (Figure 4c). Calcium-bound P also contributed to a high proportion in the FSS (Supplemental Figure S5a). Concentrations and proportions of the Fe-P fraction were similar in the streambed and in FSS (Figure 4c; Supplemental Figure S5).

Organic matter content was markedly higher throughout the U8 streambed sediment profiles (20–30%) than in C6 and E23 (3 and 4%) (Supplemental Table S3). Streambed sediment profiles from U8 also had markedly higher water content (82–93%) than either C6 (40–64%) or E23 (31–35%) (Supplemental Table S3). No significant correlations between labile P in the active streambed sediment and corresponding RP flow-weighted concentrations were observed (data not shown).

4 | DISCUSSION

4.1 | Seasonal and between-catchment variations in P fractions in FSS and active streambed sediment

Although all catchments are dominated by clay soils and had similar FSS TP concentrations, the dominant P fraction in FSS differed between catchments. The identified differences between catchments are described below, and their probable explanations fulfil Aim 1 of the study. There was also a clear temporal variation in all catchments, especially for Fe-P, answering Aim 2.

4.1.1 | Labile P in FSS and streambed sediment

Bioavailability of different P fractions in sediments depends on environmental conditions, such as pH (Lukkari et al., 2007), redox conditions (Records et al., 2016; Sharpley, 1993), and turbulence (Withers & Jarvie, 2008). The labile P fraction ($\text{H}_2\text{O-P}$ and Fe-P) together with org-P has been termed as “potentially bioavailable P” (Kaiserli et al., 2002). For $\text{H}_2\text{O-P}$, the most bioavailable fraction is PO_4^{3-} , which is directly available for uptake by aquatic organisms (Zhou et al., 2001). However, $\text{H}_2\text{O-P}$ is likely to be of minor importance in these catchments due to its low concentrations. The Fe-P fraction is usually the largest pool of bioavailable P due to its high release potential from sediment undergoing changes in redox status (Li et al., 2016). Reducing conditions can occur in streambed sediment due to intrusion of anoxic groundwater (Krause et al., 2011) but also due to high microbial and biological activity in shallow, stagnant ditches during summer low flow (Sallade & Sims, 1997). Organically bound P can become bioavailable because it can be mineralized to RP (Reitzel et al., 2005; Schütz et al., 2017). Aluminum-bound P is considered stable in the pH range of surface waters (Stumm & Morgan, 1995). Calcium-bound P is usually contained in mineral lattices and hence is usually nonavailable or even considered inert (Lukkari et al., 2007). There is a need for further studies on the effect of environmental conditions on potential P bioavailability in FSS and streambed sediment in headwater streams because the environmental conditions affecting P release in headwater streams can differ from larger rivers as well as lakes (Lannergård et al., 2020).

In all catchments, the increases in FSS Fe-P during summer and early autumn corresponded to periods of lower specific discharge (Figure 2b; Supplemental Figure S3a). This decrease in discharge, along with higher temperatures and increased biological activity, could change the redox potential in streambed sediments, resulting in a reducing environment conducive to dissolution and in the release of ferric Fe and phosphate into the water column (Records et al., 2016; Sallade & Sims, 1997; Smolders et al., 2017). When sediment is mobilized from the streambed into the potentially more aerobic water column environment, Fe oxidation may occur converting Fe^{2+} to ferric Fe^{3+} , which can react with water column RP. Such Fe oxidation and RP sorption may occur on fluvial sediment particles, resulting in higher FSS concentrations and lower water column concentrations. The increase in FSS Fe-P is prominent in both summers, with a noticeable decrease during winter and spring months, when the specific discharge is higher (Figure 2b). Flow-weighted water column RP concentrations are also lower during summer, especially at the two smaller streams (U8 and E23), supporting this theory (Supplemental Figure S3c). Because the C6 stream is much larger than the other two streams, the flow

rarely stagnates, staying in constant motion, with potentially less frequent anoxic conditions close to the streambed sediment surface. High amounts of FSS Fe-P during low flow periods, at times where anoxic conditions are more common, does suggest an increase in bioavailable P during the growing season. However, it is unknown whether these FSS particles would remain in suspension to settle in the streambed or be transported further downstream to receiving waters, potentially contributing to an increased P pool in receiving sediments.

The Fe-P fraction was dominant in FSS in C6 (Figure 3b), where soils around the catchment outlet are mainly silty clay loam. Catchment bedrock mineralogy (Table 1) does not imply a high Fe content, and there are no available measurements of Fe content in catchment soils or sediment. However, a probable explanation for the dominance of Fe-P in C6 is the presence of Fe hydroxides/oxides in the soil and sediment. Streambed sediment in C6 is similar to the FSS in that sense, with the Fe-P fraction being the second largest fraction, although decreasing with depth throughout the sediment profile. Earlier studies have found strong correlations between Fe and P in streambed sediment (Palmer-Felgate et al., 2009) and suspended sediment (Cooper et al., 2015), suggesting that when Fe is present in a higher amount under appropriate redox conditions, a majority of P will be bound to iron hydroxides/oxides. The national soil survey data show that Fe-P content in topsoils in this region, measured after extraction with ammonium lactate, was above the national average and the second highest among 20 studied regions of Sweden (Djordjic, 2015). Cooper et al. (2015) found strong correlations between Fe and P in suspended particulate matter in streams during base flow but not during storm flow. This supports our results with higher FSS Fe-P during low flow periods and lower Fe-P concentrations during high flow periods. Furthermore, McDowell et al. (2020) reported a decrease in streambed Fe-P with sediment depth, which was also observed in our study. Phosphorus release due to Fe hydroxides/oxides dissolution with progressively more reducing sedimentary conditions at depth may explain this pattern (Parsons et al., 2017).

The low concentrations and low variation in FSS $\text{H}_2\text{O-P}$ in all catchments could be related to the low solubility of P because the pattern is similar in the streambed sediment. The FSS is in constant motion; thus, it is likely that any $\text{H}_2\text{O-P}$ will either be taken up directly by biota or further transported. Similar behavior is likely to occur for $\text{H}_2\text{O-P}$ in the active streambed sediment. Simpson et al. (2019) also found low concentrations of loosely sorbed P when studying the effects of different pretreatments on streambed sediment. McDowell et al. (2020) found low concentrations of $\text{H}_2\text{O-P}$ in streambed sediment, with Fe-P as the general dominating fraction at their sampling sites. These low $\text{H}_2\text{O-P}$ concentrations (or corresponding fractions in other fractionation schemes) seem

to be quite consistent for streambed sediment and lake sediment (Kaiserli et al., 2002; McDowell et al., 2020; Pettersson, 1998).

4.1.2 | More stable P (Al-P and Ca-P) and geological conditions

In the smallest catchment (U8), Al-P is the dominant FSS P fraction, and the Al-P concentrations were markedly higher than in C6 and E23 (Figure 3c). In contrast, U8 had lower Ca-P concentrations than C6 and E23. Bedrock in U8 consists mainly of granodiorite-granite (Table 1) that contains plagioclase feldspar, generated from Al-containing minerals. Similarly, the Al-P fraction is high in U8 streambed sediment, which supports a geological weathering contribution to the Al-rich environment. The presence of particles such as Al-silicates (clay minerals) (House et al., 1998) and especially Fe and Al hydroxides and oxides (Zhang & Huang, 2007) will make RP react and bind to mineral or particle surfaces (Withers & Jarvie, 2008). The average clay content in U8 is approximately 10% higher than in C6 and E23 (Sandström et al., 2020), with Al as a primary element in most clay minerals. This also suggests more Al-P because clay minerals in soils having Al sheets in their structure have high P sorption capacity, depending on pH conditions (Gérard, 2016), where U8 has the lowest pH (Table 2). High P sorption to Al-containing clay minerals has also been found in lake sediments (Tonello et al., 2020). The C6 bedrock is instead mainly quartz arenite, and in E23, it is mainly granite. However, all catchments have similar mineralogy consisting of quartz, feldspar, and mica (Table 1). This implies that the bedrock might have less influence on P fractions in C6 and E23 because the mineralogy does not seem to be reflected in the P fractions as prominently as in U8.

The FSS Ca-P fraction was similar across all catchments, proving that parent soil material is quite comparable. However, there was one Ca-P peak in U8 at the end of the study period. This could be related to the practice of structural liming in the catchment. The practice, which is a popular mitigation measure to reduce P losses from clay soils in Sweden, is usually performed around that time of the year (Geranmayeh, 2017). Although no recorded data are available for structural liming during the study period in U8, structural liming was used in the catchment prior to 2013 when a product containing approximately 15% Ca(OH)₂ was applied at a rate of 4 t ha⁻¹ (Malgeryd et al., 2015). In U8 streambed sediment profiles, Ca-P seemed to increase with sediment depth (Figure 4b), whereas in C6, the active streambed sediment had markedly higher Ca-P concentrations that then decreased and stayed more constant down through the core (Figure 4a). Calcium-bound P represents the most tightly bound P and is thus more

stable deeper in the streambed sediment, where less, if any, exchange with the overlaying water column occurs.

4.1.3 | Organically bound P

The U8 catchment had a markedly higher OM content in streambed sediment (Supplemental Table S3) than C6 and E23. This was not reflected in the FSS, where OM content was more similar among the catchments (Supplemental Table S1). The low OM content in the streambed sediments compared with FSS in E23 and C6 was also reflected in the org-P fractions, with lower concentrations and proportions of org-P in streambed sediment but still moderately high concentrations of FSS (Figure 4b,c; Supplemental Figure S5). In U8, with a closer match in org-P content between the streambed and FSS, there was a lot of vegetation in the stream compared with C6 and E23. This catchment is also the smallest one (Table 1) and has periods of very low flow where the stream almost dries out. This contributes to slowing down the movement of water and FSS. In the larger C6 catchment, the stream flow was almost constant, and the FSS will likely not settle in the streambed for a longer time period. The vegetation itself in U8 is probably a contributor to the higher OM content in streambed sediment but could also contribute to trapping more OM by slowing flows, allowing it to settle in the streambed sediment (Cotton et al., 2006; Verschoren et al., 2017). In contrast, vegetation cover is much poorer in the coarser E23 streambed.

In E23, org-P was the dominant fraction in FSS, where it displayed a high variation in concentration but not as much variation in distribution (Figure 3d). The high concentration variation might be attributed to the peak in org-P at the start of the study period, although concentrations remained relatively high throughout, making the variation in distribution smaller. In comparison to the other catchments, FSS in E23 have generally higher org-P concentrations. The difference between the streambed and FSS in both org-P concentrations and OM content also suggests an external source of org-P. Differing from C6 and U8, E23 had a higher stocking density, higher manure inputs, and a horse paddock with a steep slope down to the stream located just upstream the sampling point (Ulén et al., 2011). The near-stream zone by the outlet of E23 is forested, unlike the other catchments that have mainly agricultural fields surrounding the outlet. The comparatively high FSS org-P concentration in E23 could reflect timing of manure application to the fields or direct inputs from the horse paddock, which are potential hotspots for P leaching (Parvage et al., 2015). This could have led to an increase in particulate P transport to the stream and thus to a peak in FSS org-P, which was not reflected in the streambed sediment. The weak but significant correlation between org-P and Fe-P might

indicate that the org-P interacts with and contributes to the pool of bioavailable P.

4.1.4 | TP and general patterns between streambed and FSS

As stated in Section 3.3, P fractions and concentrations in FSS sampled immediately prior to streambed sediment sampling are in the same concentrations ranges as in the active streambed sediment having similar proportions of dominating fractions, especially in U8 and C6 (Figure 4). In contrast, mean values of FSS P fractions and concentrations over the whole study period did not reflect the active streambed sediment as well. Due to the marked variations in FSS P fractions and concentrations over time, a comparison with the time-aggregated mean value would not be representative. Unfortunately, due to high flow conditions, the C6 FSS sample was collected over a longer time period than in U8 and E23 and thus may not be as representative. Our study supports Palmer-Felgate et al. (2009), who found a significant correlation between TP in streambed sediment and P concentration in suspended sediment in catchments having different agricultural land use intensity, but the relationships with stream water P concentrations were not as clear.

All catchments had approximately the same median FSS TP concentrations (Figure 3f) and had similar seasonal variation. Selection of catchments was based on long-term monitoring indicative of high transport of P and SS. Interestingly, the two smallest catchments had the greatest total variation in FSS TP concentrations, even though the concentrations varied in all catchments. Notably, this study has important implications for understanding the P cycle in headwater catchments. It supports the argument of Bol et al. (2018) for a greater focus on headwaters due to their high variability in P losses (Haygarth et al., 2012). Moreover, it demonstrates that focusing solely on TP dynamics will provide an incomplete indication of P mobility and reactivity among headwaters because it neglects variation in FSS P fractions and bioavailability. The variation in FSS P fractions has been shown by increases in Fe-P during the growing season.

4.2 | Water quality and P in FSS

The observed correlations between flow-weighted PP and SS concentrations (Section 3.2; Supplemental Figure S2) support earlier work in these catchments (Sandström et al., 2020). The lack of significant relationships between SS and RP (except for in U8) and the low loads and concentrations of RP reflect the dominance of PP in these catchments.

Comparisons of water column P fractions with corresponding FSS P fractions to address Aim 3 of our study showed

a weak correlation between FSS labile P and water column RP (Supplemental Figure S4). This illustrates the complexity of P circulation in small headwater catchments, where no clear relationship to flow was visible for all catchments. The U8 catchment had two outlying sampling occasions with low discharge (high FSS labile P and low water column RP concentrations; Supplemental Figure S4) and was the only catchment with a significant relationship between RP and SS. This suggests that during high discharge, some water column RP might consist of small clay particles or colloids that will pass through a filter and be identified as RP instead of PP. Indeed, Ulén (2003) showed that the colloidal P fraction was important in surface runoff, tile-drainage water, and stream water in a clay-dominated catchment in central Sweden. The significant but weak correlation between labile P and water column RP could also indicate the importance of external sources, such as manure.

In U8, where redox-independent Al-P was the dominant P fraction, FSS TP concentrations seemed to display an opposite pattern to water column TP concentrations (Supplemental Figures S3b and S5d). Here also FSS TP concentrations seemed to be highest during low flow periods, whereas water column TP was also low. This could be explained by low SS losses during low flow periods, resulting in P enrichment on SS particles (Sandström et al., 2020). It also points toward an exchange between FSS and the water column, where P could theoretically be sorbed to suspended sediment particles, increasing the concentration on the particles during periods with low concentrations of water column TP.

During the second study year there was a drought in Sweden (SMHI, 2018), resulting in low flows in all catchments (Supplemental Figure S3a), especially E23 (Supplemental Table S1). The drought affected TP loads and concentrations, with lower levels than usual in all catchments. Although the effect of the drought on FSS P fractions is difficult to establish from the 2-yr study period, effects of flow on redox conditions and temporal variability in the Fe-P fraction are visible.

5 | CONCLUSIONS

Based on time-integrated sampling of FSS in the three headwater catchments covering different seasons in combination with water quality monitoring data and streambed sediment sampling, the following conclusions can be drawn:

- There is a significant difference in the distribution of FSS P fractions in the three catchments. The differences are likely related to differences in geology, soil clay content, external P sources and flow conditions.
- There are seasonal variations, especially for Fe-P, in FSS P fractions and concentrations in all catchments that may

be due to seasonal changes in flow, temperature and redox conditions.

- Water column TP concentrations are not clearly related to FSS TP, with an almost opposite temporal pattern. There is a weak but significant positive correlation between water column RP and FSS labile P.
- The P fractions found in FSS are in the same concentration ranges as P fractions in the streambed sediment; however, the distribution of different P fractions differs. Although the dominant P fractions were the same in FSS and streambed sediment for C6 (Ca-P) and U8 (org-P, Al-P, and Fe-P), the fractional distribution in streambed sediment and FSS show no resemblance in E23.

This seasonality of FSS P fractions shows the importance of measuring different P fractions and not only TP during ecologically sensitive periods. Because the FSS may act as the interface between the streambed sediment and the water column, measuring and reporting P fractions in FSS will lead to a better understanding of the biogeochemical processes controlling P dynamics in agricultural headwaters.

ACKNOWLEDGMENTS

This study was funded by FORMAS (Grant 2017-00017) as a part of a joint EU-project funded by Water JPI. Thanks to Helena Linefur for help with reports, manure data and water quality data, Stefan Andersson for manure data and Maria Blomberg for field sampling and flow data. EEL was funded by FORMAS (Grant 2015-1518) and JR by the Biowater Nordic Center of Excellence funded by Nordforsk (project number 82263). We thank Mikael Östlund and Jasmina Sargac for assistance with field sampling and Göran Gullberg for sampling equipment and the two anonymous reviewers and the editor for their constructive comments, helping to improve the article.

AUTHOR CONTRIBUTIONS

Sara Sandström, Conceptualization, Data curation, Formal analysis, Investigation, Methodology, Project administration, Validation, Visualization, Writing-original draft, Writing-review & editing; Martyn N. Futter, Conceptualization, Data curation, Methodology, Supervision, Writing-review & editing; David W. O'Connell, Conceptualization, Funding acquisition, Methodology, Writing-review & editing; Emma E. Lannergård, Formal analysis, Investigation, Writing-review & editing; Jelena Rakovic, Formal analysis, Writing-review & editing; Katarina Kyllmar, Conceptualization, Methodology, Supervision, Writing-review & editing; Laurence W. Gill, Funding acquisition, Writing-review & editing; Faruk Djodjic, Conceptualization, Data curation, Investigation, Methodology, Project administration, Supervision, Writing-review & editing.

CONFLICT OF INTEREST

No conflicts of interest are relevant for this study.

ORCID

Sara Sandström  <https://orcid.org/0000-0003-0301-0776>
 Jelena Rakovic  <https://orcid.org/0000-0002-7653-6747>
 Faruk Djodjic  <https://orcid.org/0000-0002-2172-242X>

REFERENCES

- Agstam-Norlin, O., Lannergård, E., Futter, M., & Huser, B. (2020). Optimization of aluminum treatment efficiency to control internal phosphorus loading in eutrophic lakes. *Water Research*, 116150.
- Alexandersson, H., Karlström, C., & Larsson-McCann, S. (2001). *Temperaturen och nederbörden i Sverige 1961–90. Referensnormaler*. Sveriges Meteorologiska och Hydrologiska Institut.
- Audette, Y., O'Halloran, I. P., Nowell, P. M., Dyer, R., Kelly, R., & Voroney, R. P. (2018). Speciation of phosphorus from agricultural Muck soils to stream and lake sediments. *Journal of Environmental Quality*, 47, 884–892. <https://doi.org/10.2134/jeq2018.02.0068>
- Ballantine, D. J., Walling, D. E., Collins, A. L., & Leeks, G. J. L. (2008). The phosphorus content of Fluvial suspended sediment in three low-land groundwater-dominated catchments. *Journal of Hydrology*, 357, 140–151. <https://doi.org/10.1016/j.jhydrol.2008.05.011>
- Bol, R., Gruau, G., Mellander, P.-E., Dupas, R., Bechmann, M., Skarbøvik, E., Bierozza, M., Djodjic, F., Glendell, M., Jordan, P., van der Grift, B., Rode, M., Smolders, E., Verbeeck, M., Gu, S., Klumpp, E., Pohle, I., Fresne, M., & Gascuel-Oudou, C. (2018). Challenges of reducing phosphorus based water eutrophication in the agricultural landscapes of northwest Europe. *Frontiers in Marine Science*, 5. <https://doi.org/10.3389/fmars.2018.00276>
- Condon, L. M., & Newman, S. (2011). Revisiting the fundamentals of phosphorus fractionation of sediments and soils. *Journal of Soils and Sediments*, 11, 830–840. <https://doi.org/10.1007/s11368-011-0363-2>
- Cooper, R. J., Rawlins, B. G., Krueger, T., Lézé, B., Hiscock, K. M., & Pedentchouk, N. (2015). Contrasting controls on the phosphorus concentration of suspended particulate matter under baseflow and storm event conditions in agricultural headwater streams. *Science of the Total Environment*, 533, 49–59. <https://doi.org/10.1016/j.scitotenv.2015.06.113>
- Correll, D. L. (1998). The role of phosphorus in the eutrophication of receiving waters: A review. *Journal of Environmental Quality*, 27, 261–266. <https://doi.org/10.2134/jeq1998.00472425002700020004x>
- Cotton, J., Wharton, G., Bass, J., Heppell, C., & Wotton, R. (2006). The effects of seasonal changes to in-stream vegetation cover on patterns of flow and accumulation of sediment. *Geomorphology*, 77, 320–334. <https://doi.org/10.1016/j.geomorph.2006.01.010>
- Djodjic, F. (2015). Soil distribution and nutrient status in Swedish arable land: Summary of results from the Swedish Board of Agriculture national soil survey. *SLU, Department of Aquatic Sciences and Assessment: Report*, 2015, 11.
- Ezzati, G., Fenton, O., Healy, M. G., Christianson, L., Feyereisen, G., Thornton, S., Chen, Q., Fan, B., Ding, J., & Daly, K. (2020). Impact of P inputs on source-sink P dynamics of sediment along an agricultural ditch network. *Journal of Environmental Management*, 257, 109988. <https://doi.org/10.1016/j.jenvman.2019.109988>
- Geranmayeh, P. (2017). *Strukturkalkning i stor skala - Vad krävs och vad kostar det?* Underlag till vitbok för projektet Levande Kust. (In Swedish.) BalticSea2020.

- Gérard, F. (2016). Clay minerals, iron/aluminum oxides, and their contribution to phosphate sorption in soils: A myth revisited. *Geoderma*, 262, 213–226. <https://doi.org/10.1016/j.geoderma.2015.08.036>
- Goedkoop, W., & Pettersson, K. (2000). Seasonal changes in sediment phosphorus forms in relation to sedimentation and benthic bacterial biomass in Lake Erken. *Hydrobiologia*, 431, 41–50. <https://doi.org/10.1023/A:1004050204587>
- Haygarth, P., Page, T., Beven, K., Freer, J., Joyner, A., Butler, P., Wood, G., & Owens, P. (2012). Scaling up the phosphorus signal from soil hillslopes to headwater catchments. *Freshwater Biology*, 57, 7–25. <https://doi.org/10.1111/j.1365-2427.2012.02748.x>
- Haygarth, P. M., & Sharpley, A. (2000). Terminology for phosphorus transfer. *Journal of Environmental Quality*, 29, 10–15. <https://doi.org/10.2134/jeq2000.00472425002900010002x>
- House, W., Jickells, T., Edwards, A., Praska, K., & Denison, F. (1998). Reactions of phosphorus with sediments in fresh and marine waters. *Soil Use and Management*, 14, 139–146. <https://doi.org/10.1111/j.1475-2743.1998.tb00632.x>
- Hupfer, M., Gächter, R., & Giovanoli, R. (1995). Transformation of phosphorus species in settling seston and during early sediment diagenesis. *Aquatic Sciences-Research Across Boundaries*, 57(4), 305–324. <https://doi.org/10.1007/BF00878395>
- Hupfer, M., Zak, D., Rossberg, R., Herzog, C., & Pothig, R. (2009). Evaluation of a well-established sequential phosphorus fractionation technique for use in Calcite-Rich Lake sediments: identification and prevention of artifacts due to apatite formation. *Limnology and Oceanography-Methods*, 7, 399–410. <https://doi.org/10.4319/lom.2009.7.399>
- Huser, B. J., & Rydin, E. (2005). Phosphorus inactivation by aluminum in lakes Gårdsjön and Härsvatten sediment during the industrial acidification period in Sweden. *Canadian Journal of Fisheries and Aquatic Sciences*, 62, 1702–1709. <https://doi.org/10.1139/f05-083>
- Huser, B. J., Egemose, S., Harper, H., Hupfer, M., Jensen, H., Pilgrim, K. M., Reitzel, K., Rydin, E., & Futter, M. (2016). Longevity and effectiveness of aluminum addition to reduce sediment phosphorus release and restore lake water quality. *Water Research*, 97, 122–132. <https://doi.org/10.1016/j.watres.2015.06.051>
- Håkanson, L., & Jansson, M. (1983). *Principles of lake sedimentology*. Springer.
- Jarvie, H. P., Jürgens, M. D., Williams, R. J., Neal, C., Davies, J. J., Barrett, C., & White, J. (2005). Role of river bed sediments as sources and sinks of phosphorus across two major eutrophic UK river basins: The Hampshire Avon and Herefordshire Wye. *Journal of Hydrology*, 304, 51–74. <https://doi.org/10.1016/j.jhydrol.2004.10.002>
- Kaiserli, A., Voutsas, D., & Samara, C. (2002). Phosphorus fractionation in lake sediments: Lakes Volvi and Koronia, N. Greece. *Chemosphere*, 46, 1147–1155. [https://doi.org/10.1016/S0045-6535\(01\)00242-9](https://doi.org/10.1016/S0045-6535(01)00242-9)
- Kleinman, P. J. A., Sharpley, A. N., Buda, A. R., McDowell, R. W., & Allen, A. L. (2011). Soil controls of phosphorus in runoff: Management barriers and opportunities. *Canadian Journal of Soil Science*, 91, 329–338. <https://doi.org/10.4141/cjss09106>
- Krause, S., Hannah, D. M., Fleckenstein, J., Heppell, C., Kaeser, D., Pickup, R., Pinay, G., Robertson, A. L., & Wood, P. J. (2011). Interdisciplinary perspectives on processes in the hyporheic zone. *Ecology*, 4, 481–499. <https://doi.org/10.1002/eco.176>
- Kyllmar, K., Forsberg, L. S., Andersson, S., & Martensson, K. (2014). Small agricultural monitoring catchments in Sweden representing environmental impact. *Agriculture Ecosystems & Environment*, 198, 25–35.
- Lannergård, E. E., Agstam-Norlin, O., Huser, B. J., Sandström, S., Rakovic, J., & Futter, M. N. (2020). New insights into legacy phosphorus from fractionation of streambed sediment. *Journal of Geophysical Research: Biogeosciences*, 125, e2020JG005763.
- Li, Z., Sheng, Y., Yang, J., & Burton, E. D. (2016). Phosphorus release from coastal sediments: Impacts of the oxidation-reduction potential and sulfide. *Marine Pollution Bulletin*, 113, 176–181. <https://doi.org/10.1016/j.marpolbul.2016.09.007>
- Linefur, H., Forsberg, S. L., Kyllmar, K., Andersson, S., Johansson, G., & Blomberg, M. (2017). *Växtnäringsförluster I Små Jordbruksdominerade Avrinningsområden 2015/2016*. Institutionen för mark och miljö, Sveriges lantbruksuniversitet.
- Linefur, H., Norberg, L., Kyllmar, K., Andersson, S., & Blomberg, M. (2019). *Växtnäringsförluster I Små Jordbruksdominerade Avrinningsområden 2017/2018*. Institutionen för mark och miljö, Sveriges lantbruksuniversitet.
- Lukkari, K., Hartikainen, H., & Leivuori, M. (2007). Fractionation of sediment phosphorus revisited. I: fractionation steps and their biogeochemical basis. *Limnology and Oceanography: Methods*, 5, 433–444.
- Malgeryd, J., Stjernman-Forsberg, L., Kyllmar, K., Heeb, A., Gustafsson, J., & Svensson, A. (2015). Åtgärder mot Fosforförluster Från Jordbruksmark—Erfarenheter Från Tre Avrinningsområden I Västmanland, Östergötland Och Halland. Slutrapport Och Delrapport 2 Från Projekt Greppa Fosfor, 2010–2014. (In Swedish.) *Jordbruksverket Rapport*, 2015, 2.
- McDowell, R. W., Depree, C., & Stenger, R. (2020). Likely controls on dissolved reactive phosphorus concentrations in baseflow of an agricultural stream. *Journal of Soils and Sediments*, 20, 3254–3265. <https://doi.org/10.1007/s11368-020-02644-w>
- Murphy, J., & Riley, J. P. (1962). A modified single solution method for the determination of phosphate in natural waters. *Analytica Chimica Acta*, 27, 31–36. [https://doi.org/10.1016/S0003-2670\(00\)88444-5](https://doi.org/10.1016/S0003-2670(00)88444-5)
- Nguyen, L., & Sukias, J. (2002). Phosphorus fractions and retention in drainage ditch sediments receiving surface runoff and subsurface drainage from agricultural catchments in the North Island, New Zealand. *Agriculture, Ecosystems & Environment*, 92, 49–69.
- O’Connell, D., Ansems, N., Kukkadapu, R., Jaisi, D., Orihel, D., Cade-Menun, B., Hu, Y., Wiklund, J., Hall, R., & Chessell, H. (2020). Changes in sedimentary phosphorus burial following artificial eutrophication of lake 227, experimental lakes area, Ontario, Canada. *Journal of Geophysical Research: Biogeosciences*, 125(8), e2020JG005713.
- Owens, P. N., & Walling, D. E. (2002). The phosphorus content of Fluvial sediment in rural and industrialized river basins. *Water Research*, 36, 685–701. [https://doi.org/10.1016/S0043-1354\(01\)00247-0](https://doi.org/10.1016/S0043-1354(01)00247-0)
- Palmer-Felgate, E. J., Jarvie, H. P., Withers, P. J., Mortimer, R. J., & Krom, M. D. (2009). Stream-bed phosphorus in paired catchments with different agricultural land use intensity. *Agriculture, Ecosystems & Environment*, 134, 53–66.
- Parsons, C. T., Rezaeehad, F., O’Connell, D. W., & Van Cappellen, P. (2017). Sediment phosphorus speciation and mobility under dynamic redox conditions. *Biogeosciences*, 14, 3585–3602.
- Parvage, M. M., Ulén, B., & Kirchmann, H. (2015). Are horse paddocks threatening water quality through excess loading of nutrients? *Journal of Environmental Management*, 147, 306–313. <https://doi.org/10.1016/j.jenvman.2014.09.019>

- Pettersson, K. (1998). Mechanisms for internal loading of phosphorus in lakes. *Hydrobiologia*, 374, 21–25. <https://doi.org/10.1023/A:1017011420035>
- Phillips, J. M., Russell, M. A., & Walling, D. E. (2000). Time-integrated sampling of fluvial suspended sediment: a simple methodology for small catchments. *Hydrological Processes*, 14, 2589–2602. [https://doi.org/10.1002/1099-1085\(20001015\)14:14%3c2589::AID-HYP94%3e3.0.CO;2-D](https://doi.org/10.1002/1099-1085(20001015)14:14%3c2589::AID-HYP94%3e3.0.CO;2-D)
- Psenner, R. (1988) Fractionation of phosphorus in suspended matter and sediment. *Archiv für Hydrobiologie. Beihefte. Ergebnisse der Limnologie*, 30, 98–110.
- Psenner, R., & Pucsko, R. (1988). Phosphorus fractionation: Advantages and limits of the method for study of sediment P origins and interactions. *Archiv für Hydrobiologie Beiheft—Ergebnisse der Limnologie*, 30, 43–59.
- Psenner, R., Pucsko, R., & Sager, M. (1984). Die fractionierung organischer und anorganischer Phosphorverbindungen von Sedimenten. *Archiv für Hydrobiologie—Supplementband*, 70, 111–155.
- R Core Team. (2019). *R: A language and environment for statistical computing*. <https://www.R-project.org/>
- Records, R. M., Wohl, E., & Arabi, M. (2016). Phosphorus in the river corridor. *Earth-Science Reviews*, 158, 65–88. <https://doi.org/10.1016/j.earscirev.2016.04.010>
- Reitzel, K., Ahlgren, J., Gogoll, A., Jensen, H. S., & Rydin, E. (2006). Characterization of phosphorus in sequential extracts from lake sediments using ^{31}P nuclear magnetic resonance spectroscopy. *Canadian Journal of Fisheries and Aquatic Sciences*, 63, 1686–1699. <https://doi.org/10.1139/f06-070>
- Reitzel, K., Hansen, J., Andersen, F. Ø., Hansen, K. S., & Jensen, H. S. (2005). Lake restoration by dosing aluminum relative to mobile phosphorus in the sediment. *Environmental Science & Technology*, 39, 4134–4140.
- Reynolds, C., & Davies, P. (2001). Sources and bioavailability of phosphorus fractions in freshwaters: A British perspective. *Biological Reviews*, 76, 27–64. <https://doi.org/10.1017/S1464793100005625>
- Ruttenberg, K., Ogawa, N., Tamburini, F., Briggs, R., Colasacco, N., & Joyce, E. (2009). Improved, high-throughput approach for phosphorus speciation in natural sediments via the sedex sequential extraction method. *Limnology and Oceanography: Methods*, 7, 319–333.
- Sallade, Y., & Sims, J. (1997). Phosphorus transformations in the sediments of Delaware's agricultural drainageways: II. Effect of reducing conditions on phosphorus release. *Journal of Environmental Quality*, 26, 1579–1588. <https://doi.org/10.2134/jeq1997.00472425002600060018x>
- SanClements, M. D., Fernandez, I. J., & Norton, S. A. (2009). Soil and sediment phosphorus fractions in a forested watershed at Acadia National Park, ME, USA. *Forest Ecology and Management*, 258, 2318–2325.
- Sandström, S., Futter, M. N., Kyllmar, K., Bishop, K., O'Connell, D. W., & Djodjic, F. (2020). Particulate phosphorus and suspended solids losses from small agricultural catchments: Links to stream and catchment characteristics. *Science of the Total Environment*, 711, 134616. <https://doi.org/10.1016/j.scitotenv.2019.134616>
- Schindler, D. W. (1974). Eutrophication and recovery in experimental lakes: Implications for lake management. *Science*, 184, 897–899. <https://doi.org/10.1126/science.184.4139.897>
- Schütz, J., Rydin, E., & Huser, B. J. (2017). A newly developed injection method for aluminum treatment in Eutrophic Lakes: Effects on water quality and phosphorus binding efficiency. *Lake and Reservoir Management*, 33, 152–162.
- SGU. (2016). *Product: Bedrock 1:50 000 - 1:250 000*. Geological Survey of Sweden.
- Sharpley, A. N. (1993). An innovative approach to estimate bioavailable phosphorus in agricultural runoff using iron oxide-impregnated paper. *Journal of Environmental Quality*, 22, 597–601. <https://doi.org/10.2134/jeq1993.00472425002200030026x>
- Sharpley, A. N., Kleinman, P. J. A., Jordan, P., Bergstrom, L., & Allen, A. L. (2009). Evaluating the success of phosphorus management from field to watershed. *Journal of Environmental Quality*, 38, 1981–1988. <https://doi.org/10.2134/jeq2008.0056>
- Sharpley, A., Jarvie, H. P., Buda, A., May, L., Spears, B., & Kleinman, P. (2013). Phosphorus legacy: Overcoming the effects of past management practices to mitigate future water quality impairment. *Journal of Environmental Quality*, 42, 1308–1326. <https://doi.org/10.2134/jeq2013.03.0098>
- Simpson, Z. P., McDowell, R. W., & Condon, L. M. (2019). The error in stream sediment phosphorus fractionation and sorption properties effected by drying pretreatments. *Journal of Soils and Sediments*, 19, 1587–1597. <https://doi.org/10.1007/s11368-018-2180-3>
- Sveriges Meteorologiska och Hydrologiska Institut (SMHI). (2018). *Sommaren 2018—Extremt varm och solig*. SMHI.
- Smolders, E., Baetens, E., Verbeeck, M., Nawara, S., Diels, J., Verdier, M., Peeters, B., De Cooman, W., & Baken, S. (2017). Internal loading and redox cycling of sediment iron explain reactive phosphorus concentrations in lowland rivers. *Environmental Science & Technology*, 51, 2584–2592.
- Söderström, M., & Piikki, K. (2016). *Digital soil map: Detailed mapping of soil texture in the topsoil of the arable land*. (In Swedish.) Swedish University of Agricultural Sciences, Department of Soil and Environment, Precisionsodling och pedometri, Precisionsodling.
- Stumm, W., & Morgan, J. (1995). *Aquatic chemistry: Chemical equilibria and rates in natural waters* (3rd ed.). Wiley.
- Tonello, M. S., Hebner, T. S., Sterner, R. W., Brovold, S., Tiecher, T., Bortoluzzi, E. C., & Merten, G. H. (2020). Geochemistry and mineralogy of Southwestern lake superior sediments with an emphasis on phosphorus lability. *Journal of Soils and Sediments*, 20, 1060–1073. <https://doi.org/10.1007/s11368-019-02420-5>
- Ulén, B. (2003). Concentrations and transport of different forms of phosphorus during snowmelt runoff from an Illite clay soil. *Hydrological Processes*, 17, 747–758. <https://doi.org/10.1002/hyp.1164>
- Ulén, B., Djodjic, F., Etana, A., Johansson, G., & Lindström, J. (2011). The need for an improved risk index for phosphorus losses to water from tile-drained agricultural land. *Journal of Hydrology*, 400, 234–243. <https://doi.org/10.1016/j.jhydrol.2011.01.038>
- Verschoren, V., Schoelynck, J., Cox, T., Schoutens, K., Temmerman, S., & Meire, P. (2017). Opposing effects of aquatic vegetation on hydraulic functioning and transport of dissolved and organic particulate matter in a lowland river: A field experiment. *Ecological Engineering*, 105, 221–230. <https://doi.org/10.1016/j.ecoleng.2017.04.064>
- Weigelhofer, G., Ramião, J. P., Pitzl, B., Bondar-Kunze, E., & O'Keefe, J. (2018). Decoupled water-sediment interactions restrict the phosphorus buffer mechanism in agricultural streams. *Science of the Total Environment*, 628, 44–52. <https://doi.org/10.1016/j.scitotenv.2018.02.030>
- Williams, J., Jaquet, J., & Thomas, R. (1976). Forms of phosphorus in the surficial sediments of Lake Erie. *Journal of the Fisheries Board of Canada*, 33, 413–429. <https://doi.org/10.1139/f76-063>

- Withers, P., & Jarvie, H. (2008). Delivery and cycling of phosphorus in rivers: A review. *Science of the Total Environment*, *400*, 379–395. <https://doi.org/10.1016/j.scitotenv.2008.08.002>
- Ye, H. M., Yuan, X. Y., Han, L., Yin, H., & Jin, J. (2017). Comparison of phosphorus fraction distribution and influencing factors of suspended and surface sediments in the Tiaoxi Watershed, China. *Water Science and Technology*, *75*, 2108–2118. <https://doi.org/10.2166/wst.2017.079>
- Zhang, J.-Z., & Huang, X.-L. (2007). Relative importance of solid-phase phosphorus and iron on the sorption behavior of sediments. *Environmental Science & Technology*, *41*, 2789–2795.
- Zhou, Q., Gibson, C. E., & Zhu, Y. (2001). Evaluation of phosphorus bioavailability in sediments of three contrasting lakes in China and the UK. *Chemosphere*, *42*, 221–225. [https://doi.org/10.1016/S0045-6535\(00\)00129-6](https://doi.org/10.1016/S0045-6535(00)00129-6)

SUPPORTING INFORMATION

Additional supporting information may be found online in the Supporting Information section at the end of the article.

How to cite this article: Sandström, S., Futter, M. N., O'Connell, D. W., Lannergård, E. E., Rakovic, J., Kyllmar, K., Gill, L. W., & Djodjic, F. Variability in fluvial suspended and streambed sediment phosphorus fractions among small agricultural streams. *J. Environ. Qual.* 2021;50:612–626. <https://doi.org/10.1002/jeq2.20210>